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Novel Formulations of Green Synthesized Plant Based Metal Nanoprticles along with their Therapeutic Applications: an Insight to Nano-Green world

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ABSTRACT

Nanoparticles are the most important entity in the widely spread field of nanotechnology. A series of numerous physical and chemical methods are involved in nanoparticle synthesis. Metallic nanoparticle can overcome the issues of highly monodispersed nanoparticles of various sizes, geometries and chemical composition, as they are comparatively smaller in size. Hence chemicals and non-polar solvents are involved in the synthesis of metallic nanoparticles which makes them unsuitable for being used in clinical fields. Therefore, the scientists have developed new aspects of clean, non-toxic, biocompatible and eco-friendly synthesis method for nanoparticles. Metal nanoparticles have attained a special focus attributed to their unique features like size and shape dependant optical, electrical and magnetic properties. The green synthesis using biological molecules obtained from plant sources are quite beneficial over other physical and chemical methods that have been used for metal nanoparticle synthesis and stabilization. This exhaustive review is focused on the metallic nanoparticles, which are synthesized from plant sources and overview of their pharmacological properties. **Key words:** Size and shape dependant optical, electrical and magnetic properties, Green synthesis.

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INTRODUCTION

FDA approves number of chemically synthesized newer molecules nowadays, which are introduced in the market with wide therapeutic efficacy, but the adverse effects related to these molecules can be harmful for the patients. Due to peak and valley fluctuations, conventional therapy is non- targetable in tissues and organs and high dose frequency is also the main problem associated with allopathic medicines which lead to poor patient compliance [1].

A number of phytoconstituents belonging to nature have different biological activities against chronic diseases and have wide therapeutic potencies. Phytoconstituents are beneficial as they exhibits free from adverse effects treatment where none of the medication can do. Although, some physiochemical factors like less solubility, less permeation and non-targeting at the active site will act as a barrier which create problems to its therapeutic efficacy. Therefore, various novel formulation techniques are discovered to overcome these barriers and achieve uniform drug targeting at the active site in desired concentration and enhanced therapeutic potency. These novel formulation techniques includes emulsion-based formulations, phytosomes, liposomes, microspheres, topical based formulations and nanoparticles which are available in commercial level to enhance the bioavailability of the poorly soluble herbal drug [2]. From the past few decades novel drug delivery system (NDDS) is used and gained the attention related to further development in these novel systems. The two ideal requirements for a system to be novel are:-

- Drug delivery at a predetermined rate and for predetermined span of time;
- > Conveying the active entity to the target site.

Currently, there is no such system which can fulfill all these requirements. So a lot of research work is required to accomplish them using novel strategies. These targets can be achieved by studying the drug distribution through unifying drug into a carrier system, modification in molecular drug composition or by controlling drug release in the bioenvironment to achieve desired distribution profile. Novel drug delivery systems can be effectively minimize the side effects and maintain uniform and potent levels of drug in the body. These carriers have the ability to restrict the drug action specifically in diseased tissue or organ or adjacent to it [3].

NANOPARTICLES

The Greek term 'nano' means small in size which is a unit prefix meaning "one billionth" from the range 9 to 10. Particles which have two or more dimensions in the size range as 1 to 100 NM are defined as nanoparticles [4]. A Professor from Tokyo Science University, Dr. Norio Taniguchi was introduced the term Nanotechnology in the year 1974 to explain precise production of materials at the nanometer level [5]. Although a physicist, Professor Richard P. Feynman was the first who use this Nanotechnology concept in his lecture entitled "There's plenty of room at the Bottom" [6]. Nanoparticles are colloidal particles having submicron size ranges from 10 to 1000 nm. Nanospheres are a matrix type of structure which entrapped active pharmaceutical ingredient. Nanocapsules composed from a polymer membrane which contains the API in its polymer core. Nanoparticles are the most suitable delivery tools for encapsulating drugs ranging from small to larger molecular weight compounds. Nanoparticles based on herbal drugs, is the area of thirst in modern era. Nanoparticles can release the drug through bulk erosion from the matrix or through surface erosion within the polymer, which is totally related to the nature of drug and preparation method used [7]. Nanotechnology is very popular nowadays, because of its wide range of applications from cosmetics and skin care products to abrasives, car polishes and drug delivery systems. Nanoparticles made by human are fabricated within the diameter range of < 100 nm, which shows specific physicochemical properties attributed to their small size measurement, large surface-tovolume ratio and also in-creased reactivity [8]. These specific properties of nanoparticles are responsible for the desired result which is given by an almost inert material at nanoscale [9].

USE OF NANOPARTICLES

Nanomaterials have already been exhausted deeply in early 2005 for its capabilities to use in medical and pharmaceutical fields [10]. Nanoparticle has a wide area of applications in energy, nutrition and medicine [11]. Nanotechnology has showed great possibilities nowadays in various areas of technology and science. Pharmaceutical nanotechnology includes various benefits which increasingly gained the attention of a number of budding researchers [12]. The importance of nanomaterials as drug delivery systems has been analyzed from about past twenty years which results in dosage forms with enhanced therapeutic effects as well as improved physicochemical properties [13 and 14]. Therefore, nanoparticles are already recognized as the biomaterials that have a great potential for medical and biological applications. They may be utilized as a contrast agent for medical imaging, or in therapeutic drug delivery, elimination of tumours and labelling of cells. Moreover, biomedical instruments are mostly fabricated by organic and inorganic nanoparticles in the industry, attributed to their easy incorporation in biological processes [9]. A size dependant physicochemical characteristic of nanoparticles is the reason behind its extensive exploration in the area of medicine. The size of nanoparticles and most biological molecules and structures are very similar to each other. This unique property makes them suitable candidate to apply in both *in vivo* and *in vitro* biomedical research. Due to their amalgamation in medicinal field they can be easily applied in imaging, sensing, artificial implants and targeted drug delivery. Nanoparticle used as antimicrobial is another interesting approach for their exploration in medicinal field to target highly pathogenic and drug resistant microbes. However, nanoparticle application in biology is greatly depends upon biocompatibility. Biocompatibility is the ability of material to give therapeutic effect without showing any unwanted local or systemic effects [15].

METAL BASED NANOPARTICLES

From the few past decades, metals are used for treating a number of infectious diseases and due to the emergence of resilient pathogens their antimicrobial potencies are being reevaluated nowadays. The main area of research related to metal efficacy includes their use in topical/therapeutic as well as disinfecting agent to control the bacterial adhesion and transmission. Nanoparticles can be a fascinating candidate in that because of their formulation with a high surface area to volume ratio and with unexpected morphological characters containing sharp edges and corners [16]. The metal nanoparticles have been attracted many researchers and scientists due to their widely uses in industry and medicine [17 and 18]. Antibacterial nanomaterials as compared to antibacterials are become very popular because they have the ability to give better options against antibiotics. They also overcome the problems related to antibiotics. They attributed to combat multidrug-resistant mutants and biofilms of the bacteria [19 and 20].

Metals and their oxides are exhaustively analyzed and reported for having the antibacterial potency [21]. Metals like, zinc (Zn), silver (Ag), copper (Cu), gold (Au) and titanium (Ti) possesses good antibacterial ability and have been used from ancient times [22]. Similarly, the oxides of metals as nanoparticles like iron oxide (Fe3O4), zinc oxide (ZnO), silver (Ag), titanium oxide (TiO2) and copper oxide (CuO) were

reported for having potent antibacterial activity [23]. Figure 1 shows various metals used for the synthesis of plant mediated nanoparticles.

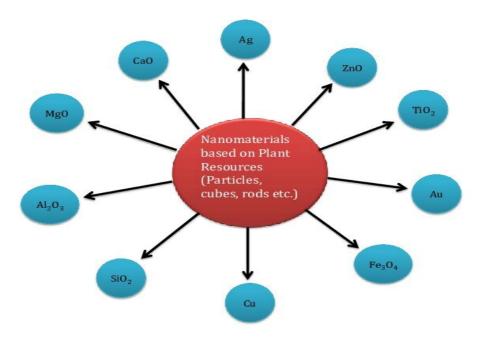


FIGURE1: VARIOUS METAL NANOPARTICLES SYNTHESIZED FROM PLANT RESOURCES [24].

SILVER (Ag) and SILVER OXIDE (Ag₂O) NANOPARTICLES

Silver nanoparticles are most popular, among all metal nanomaterials, as potent antimicrobial agent against various bacteria, fungi, and viruses [25]. According to previous literature, the antimicrobial ability of silver nanoparticles is size-dependent, similar to other matal nanoparticles [26]. Previous literatures show that the antibacterial action of silver nanoparticles is attributed to the damage of the outer membrane of bacteria [27]. However few researchers pretend that, silver nanoparticles can create pits and gaps in the bacterial membrane which causes fragmentation of the cell [28 and 29]. The mechanism behind bacterial cell death by silver is that silver ions interact with disulfide or sulfhydryl groups of enzymes results in the interruption in metabolic processes and finally causes the cell death [30]. Jo et al. studied the effect of size reduction on the antimicrobial efficacy of silver nanoparticles. Silver nanoparticles sizes ranging from 20 to 30 nm shows better penetration and can colonize within the plant tissue. According to them, silver nanoparticles can be a great option for inhibition of spore-producing fungal plant pathogens. They give silver nanoparticles more preference over synthetic fungicides in terms of toxicity [31]. Pal et al. compared the shape dependent antibacterial efficacy between spherical, rodshaped and truncated triangular shaped silver nanoparticles. They concluded that truncated triangular nanoparticles were highly reactive due to their high-atom-density surfaces and hence showed relatively high antimicrobial potency [32].

Silver oxide (Ag₂O) nanoparticles have also been detected for their higher antimicrobial activity [33]. Sondi et al. studied that *E. coli* DNA lost its replication ability when exposed to silver oxide nanoparticles and their cell cycle ceased at the G2/M phase results in DNA damage. Then the cells were affected by oxidative stress, and apoptosis was induced [34].

ZINC (Zn) and ZINC OXIDE (ZnO) NANOPARTICLES

The Zn nanoparticles functionalized with curcumin showed excellent inhibition activity against the microbial strains tested over all [35]. Aqueous extract of *Panax ginseng* roots were used to synthesize zinc nanoparticles, which is considered as the first test that results in remarkable growth reduction of cancer cells *in vitro*, by using zinc nanoparticles of Ginseng (*Panax ginseng*) [36].

ZnO nanoparticles are well known for their antibacterial efficacy against a large variety of microbes and according to reports their activity is attributed to the selected concentrations and particle size [37]. Zinc oxide nanoprticles are comparatively economical [38] and having size dependent efficacy [37] against various species of microorganisms [39 and 40]. These include pathogens such as *Salmonella enteritidis, Listeria monocytogenes* [41], *Klebsiella pneumonia* [42], *Streptococcus mutans, Lactobacillus* [43], and *E. coli* [41 and 44] along with lesser cytotoxicity [45]. The properties like white color, UV-protection and

bacterial biofilm inhibition are suggested these nanoparticles to be incorporated in facbrics [46] and also in glass [47] industries for coating of medical and related devices. In addition, FDA has also approved the use of zinc as a food additive [48]. These nanoparticles show the activity through membrane binding, inhibiting their potential and integrity, and through facilitating ROS production [41, 44 and 49]. Zinc nanoparticles are also mutagens, but relatively weak ones [50].

TITANIUM OXIDE (TiO₂) NANOPARTICLES

Titanium dioxide (TiO₂) is also one of the metal oxides, which is exhaustively analyzed and shows potent antibacterial effects [33]. This metal oxide was broadly used against both Gram-positive and Gram-negative bacteria since past few decades [51]. These are photocatalytic as they are toxic which is facilitated by visible light, near-UV or UV [19], stimulates ROS burst. The ROS damage the membrane, DNA and many other macromolecules and functions of the bacterial cell [48]. Titanium dioxide is very effective against various highly resistant microorganisms such as bacterial spores of *Bacillus* [52]. The conjugation of titanium or its oxide to other nanomaterials for example silver, possibly shows increased antibacterial activity due to synergism [53, 54 and 55].

GOLD (Au) NANOPARTICLES

Gold nanoparticles act as an antibacterial through attachment to the bacterial membrane, then alter the membrane potential and decreases ATP level, resulting inhibition of tRNA binding to the ribosome [56]. A previous literature shows the efficacy of gold and silver nanoparticles against *E. coli* and *Bacillus Calmette-Guérin* (BCG). They claimed that, Au and Ag nanoparticles exhibited significant antibacterial activity against both Gram negative (E. coli) and the Gram positive bacteria (*Bacillus Calmette-Guérin*). They also formulated gold nanoparticles by using a firmly bound capping agent, poly-allylamine hydrochloride as well as by an infirm bound capping agent, citrate. Due to its positively charged nature, poly-allylamine hydrochloride can make direct contact with the bacterial cell membrane [57 and 58]. Some other studies showed that the gold nanoparticles (5nm) can reduce 90–95% colonies of *Salmonella typhi* and *E. coli*. The investigators finally conclude that, the roughness and the dispersion of the gold nanoparticles on the medium are the main factors which affect their biocidal properties [59].

IRON OXIDE (Fe₃O₄) NANOPARTICLES

Iron oxide is an antibacterial in nano range but it is an inert material in bulk form. Previous microbiological assays conclude the antiadherent properties of iron oxide nanoparticles and its activity against both Gram-negative and Gram-positive bacteria [60]. Gold nanoparticle, conjugated with iron oxide is a newer technique against microbes which is induced by photothermal treatment [61].

COPPER (Cu) and COPPER OXIDE (CuO) NANOPARTICLES

Copper (Cu) nanoparticles due to their unique biological, chemical and physical properties, antimicrobial activities as well as the low cost of preparation, are very fascinating to the investigators [62, 63 and 64]. Usman et al. analysed the antimicrobial potential of Copper-chitosan nanoparticles (2–350 nm). Antifungal and antibacterial efficacies of these nanoparticles were studied on various microorganisms, such as methicillin resistant *S. aureus, Salmonella choleraesuis, B. subtilis, C. albicans* and *P. aeruginosa*. These results show the antimicrobial potency of these nanoparticles [63]. However, on exposure to the air, copper nanoparticles rapidly get oxidation which hinders their application [63 and 65]. Copper (Cu) is comparatively economical than other nano ranged metals and therefore it can be used for increasing efficacy in the form of nanocomposites [23].

Mahapatra et al. analysed the antibacterial potential of copper oxide nanoparticles against various microorganisms like *Salmonella paratyphi, P. aeruginosa, Klebsiella pneumoniae* and *Shigella strains.* According to their report, these nanoparticles indicated suitable antibacterial activity against the mentioned bacteria. They suggested that these nanoparticles can cross through the bacterial cell membrane and causes damage to the vital enzymes and finally trigger cell death. According to them these nanoparticles were also non cytotoxic on HeLa cell line [65]. Azam et al. investigated the antibacterial activity of copper oxide nanoparticles against *B. subtilis* and *S. aureus* (Gram-positive bacteria), and *E. coli* and *Pseudomonas aeruginosa* (Gram-negative bacteria). Their tests show the results that these nanoparticles inhibit the growth of both groups of the above mentioned bacteria. The investigators suggested that bactericidal activity of copper oxide nanoparticles is size-dependent and also affected by its stability and concentration added to the growth medium. The authors believed that the metal nanoparticles can passes through the nanometric pores of the cellular membranes of most bacteria and there by restrict the bacterial growth [66]. Copper oxide (CuO) nanoparticles, like the other metallic nanoparticles, exert their antibacterial activity [67 and 68] by membrane disruption and ROS production

[19]. The bacteria like *B. anthracis and B. subtilis* were more susceptible to Cu nanoparticle [69 and 70]. *B. subtilis*, bacteria with ample proportions of amine and carboxyl groups in its cell walls, can strongly bind to CuO and therefore more sensitive to it [38, 48 and 19]. Hence copper oxide nanoparticles create the impression that, their use in special cases would be more beneficial instead of using other metal nanoparticles [23].

SILICON (Si) and SILICON DIOXIDE (SiO₂) NANOPARTICLES

According to some previous studies, nanowires of Silicon (Si) could interface between the living cells and bacteria, and there by it interrupt the bacterial cell growth, adhesion and spreading. Some investigators stated the good antibacterial activity of silver nanoparticles with silicon nanowires in their studies. A fair biocompatibility was also seen between these nanostructures and the human lung adenocarcinoma epithelial cell line A549 [71 and 72]. Fellahi et al. prepared solicon nanowire substrates, decorated with silver and copper nanoparticles and evaluate their antibacterial activity. According to the authors, their prepared nanoparticles revealed strong antibacterial activity against *E. coli*. However the results conclude that silver based silicon nanowires shows biocompatibility with human lung adenocarcinoma epithelial cell line A549, but in case of copper based silicon nanowires, it shows cytotoxicity [71].

Egger et al. studied the antimicrobial activity of novel silver–silicon nanocomposite and concluded that, as compared to conventional materials like silver zeolite and silver nitrate, this nanocomposite has strong antimicrobial capabilities against various microorganisms [73]. At nano-range, SiO₂ more significantly shows antimicrobial efficacy due to the increased surface area [74]. In another study, Mukha et al. formulated Ag/SiO₂ and Au/SiO₂ nanostructures and evaluated their antimicrobial potency. According to their results Ag/SiO₂ nanocomposites found to be a fair antimicrobial agent against *S. aureus, C. albicans* and *E. coli*, while Au/SiO₂ nanocomposites did not exhibit any antibacterial activity against the same microorganisms. Hence the investigators recommended these nanocomposites for being useful in medical and pharmaceutical field, and also for water disinfection [75].

The above studies proposed that formulation of silicon nanocomposites by using other metals like silver can be a better alternative for antimicrobial agents. Furthermore, the non-toxicity of silicon nanoparticles is a remarkable factor for being used in biomedical fields [76].

ALUMINUM OXIDE (Al₂O₃) NANOPARTICLES

To establish the ability of aluminum oxide (Al₂O₃) nanoparticles as antibacterial agent requires more research. Though, previous researches shows comparatively mild bactericidal effects of aluminum oxide and their efficacy is largely depends on their higher concentrations [19 and 77] as well as in association with other metal nanoparticles like silver [78]. One more problem associated with them is that, they are capable of promoting horizontal multi resistance gene transfer, mediated by plasmids across genera [77]. Few years ago the mechanism of action of aluminum nano material was reported aginst *E. coli*, which was based on diffusion and accumulation within the bacterial cells followed by pit formation, perforation and membrane disorganization, finally causes death of the cell [79].

MAGNESIUM OXIDE (MgO) NANOPARTICLES

Magnesium (Mg) can be used in various nanoparticles in the form of magnesium oxide (MgO) or MgX₂ [e.g., Magnesium fluoride (MgF₂)] [19 and 80]. Magnesium based nanoparticles are capable of inducing ROS and can also inhibit the essential bacterial enzymes [48].

Magnesium oxide (MgO) has been proven for its bactericidal potency against Gram-positive as well as Gram-negative bacteria [81]. Researches give the proof that, magnesium oxide nanoparticles can disrupt the bacterial cell membrane followed by leakage of intracellular constituents and finally results in cell death [82]. Some studies show that magnesium oxide triggered the alterations in sensitivity within *E. coli* promoted by active oxygen [83]. However, Leung et al. reported that magnesium oxide nanoparticles can exhibit their high antibacterial potency without occurrence of any ROS production. The authors concluded that the antimicrobial mechanism of magnesium oxide nanoparticles depends upon the destruction of bacterial cell membrane [84]. Some researchers studied the antibacterial efficacy of MgO against *E. coli* or *S. aureus*. They proposed that the antibacterial activity of MgO nanoparticles is depend on the factor that active oxygen, such as superoxide, was present on their surface [85].

CALCIUM OXIDE (CaO) NANOPARTICLES

The potent antibacterial property of calcium oxide (CaO) is associated with the alkalinity and also with active oxygen species. The antibacterial mechanism of CaO nanoparticles has been proven that it is attributed to the generation of active oxygen, such as superoxide, which is present on their surface, and additionally to the higher pH values, through the hydration of CaO with water [86]. Jeong et al.

investigated the antimicrobial efficiency of $CaCO_3$ nanoparticles. The authors indicated that, $CaCO_3$ is transformed in to CaO, through the heat treatment. These nanoparticles has good bactericidal efficacy against *B. subtilis, S. typhimurium, E. coli* and *S. aureus* [87]. The above results suggested that CaO nanoparticles alone or in combination with other disinfectants exhibit greater antibacterial properties. CaO nanoparticles are easily available, economic and biocompatible. Hence, these properties make it a promising antibacterial agent [76]. The investigators finally proposed that these nonmaterial's have applications in environmental preservation, food processing and medical treatments [88].

CLASSICAL APPROACHES FOR THE SYNTHESIS OF METAL NANOPARTICLES

Nanoparticles are generally synthesized by two approaches, either top-down or bottom-up approaches [Figure 2]. In top-down approaches, bulk materials are successively breaking down in to nanosized structures by the use of size reduction mechanical methods. Bottom- up approach is based on the assembly of atoms or molecules to molecular structure in nanoscale range [89]. In top-down approaches, nanoparticle synthesis is achieved by size reduction of primary suitable material. This reduction in size may be performed by various physical and chemical treatments. The major drawback associated with this approach, is that the surface structure of the product is found to be non uniform, as in surface chemistry and the other physical properties of nanoparticles, primarily based on the surface structure [90]. The bottom- up approach is based on nanoparticle production by assemblage of small molecules either by joining the atoms or molecules and small structures [91]. In this the nanostructure, building blocks of the nanoparticles are formulated first and then assembled to produce the final particle [90]. The chemical as well as biological methods of metal nanoparticle production are based on the bottom-up approaches [89]. Another approach for metal nanoparticle synthesis is bottom-to-top approach, which involves chemical reduction methods [91]. Bottom-to-top synthesis not only comprises of toxic chemicals but also generates environmentally hazardous byproducts. Considering all the above stated drawbacks, this method is not commonly used for metal nanoparticle synthesis. General methods used for metal nanoparticle synthesis are as follows:

PHYSICAL METHOD OF NANOPARTICLE SYNTHESIS

Physical methods of nanoparticle synthesis involve different techniques like UV irradiation, laser ablation, radiolysis, sonochemistry etc. Physical method involves vaporization of metal atoms followed by condensation on different supports and leads to rearrangement and accumulation of metal atoms to form small clumps of metal nanoparticles [91]. By applying physical approaches, highly pure and definite shaped nanoparticles are obtained. Though, highly sophisticated instruments, chemicals and radiative heating involves in physical approaches, along with high power consumption, this method becomes expensive to operate [92].

CHEMICAL METHOD OF NANOPARTICLE SYNTHESIS

This method comprises of reduction of metal ions within the solution using different chemicals as reducing agents. Small clumps of metals are formed either by nucleation or aggregation process, which is purely based on the conditions of reaction mixture. The chemicals generally used as reducing agents are hydrogen, sodium borohydride and hydrazine [93]. This method also involves various synthetic or natural polymers as stabilizing agents like chitosan, cellulose, natural rubber and co-polymers micelles. Due to the hydrophobic nature of the above chemicals, they needed the addition of some organic solvents such as ethane, toluene, dimethyl formaldehyde and chloroform. These chemicals are toxic in nature and are non-biodegradable, which limit the production scale. Further, nanoparticle surfaces get contaminated by a few toxic chemicals, which leads to their unsuitability for certain biomedical applications [94]. Therefore, due to such major drawbacks associated with physical and chemical methods, an alternative approach for metal nanoparticle synthesis is an area of interest for investigators.

BIOLOGICAL METHOD OF NANOPARTICLE SYNTHESIS

From the past few years, biological synthesis of metallic nanoparticle has attracted considerable attention. Biogenic synthesis process involves plants and microorganisms for synthesizing nanoparticles [95]. Biogenic synthesis process of nanoparticle synthesis has the advantages over other physicochemical methods of formulation, as this method gives nanoparticles with a better defined size and morphology [96]. The nanoparticles synthesized with microorganism is fruitful in context of pharmacological applications, because of its eco-friendly nature, compatibility to use and is also readily scalable, but this process is quite expensive in comparison to the production with plant-based materials. Plant-based synthesis process is more eco-friendly, low-priced and can be easily scale-up for the large-scale synthesis

of nanoparticles, therefore it is more beneficial when compared to the traditional physicochemical methods and further, this method doesn't required high temperature, pressure and toxic chemicals [97]. Biogenic production of metal nanoparticles based on microorganisms like algae, bacteria, fungi and plants were reported in plenty of research articles. This is purely attributed to their antioxidant or reducing potencies which are responsible for metal nanoparticle reduction. Further, microbial synthesis needs immense aseptic conditions with sophisticated care and therefore, this method is not suitable for large scale production. On the other hand, plant based synthesis of nanoparticle is quite beneficial over microbes based methods, due to its easy scale-up process with less maintenance of cell culture [98]. Nanoparticle synthesis with plant extract is also fruitful, because it cut down the sophistication, related to isolation of microbes and culture medium preparation, and therefore this method is highly cost-effective and practical when compared to microorganism based synthesis. Plant based synthesis method is a onestep process of synthesis, while microbial synthesis takes long time duration and sometimes unable of producing nanoparticles due to mutation; thus, research on plant is expanding rapidly [99]. Various nanoparticle synthesis methods are discovered such as metal ions chemical reduction within aqueous solutions in presence or absence of stabilizing agents, thermal decomposition with organic solutions. The advantages of green synthesis of nanoparticles over the physical and chemical methods are:

- Clean and eco-friendly approach, as toxic chemicals are not used;
- The active biological component itself act as reducing and capping agent , therefore reduction in the overall cost of synthesis process;
- > Can be used at large scale production of nanoparticles;
- External experimental conditions like high energy and high pressure are not required, which leads to energy saving process [100].

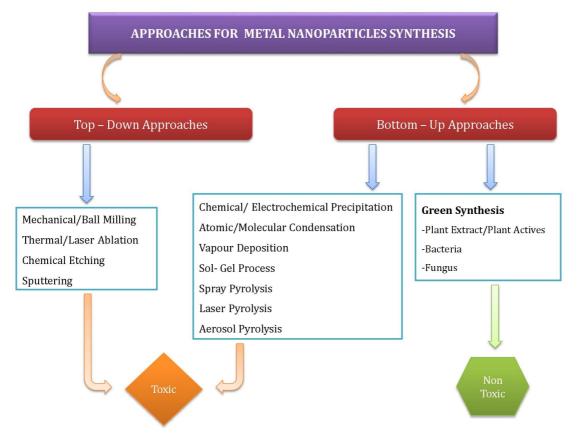


FIGURE2: VARIOUS APPROACHES OF SYNTHESIS OF METAL NANOPARTICLES {Recreated from [101]}.

PLANT EXTRACTS-MEDIATED SYNTHESIS OF NANOPARTICLES

To prepare the plant extract, different parts of the plants are used as fresh or dry material such as the fruit, leaf, peel, petal and shoot. The extraction process includes saturation of the plant material in a green solvent with or without stirring, along with subsequent filtration and centrifugation. The filtered extract is rich in the reducing and capping agents, which required for the bioreduction of metallic ions [Figure 3]. The advantage of using dried plant is that, it has a long shelf life at room temperature, but it is important

to store the fresh plant at -20° C to avoid any deterioration. The dry plant material should also be free from seasonal factors that can cause the variations in plant constituents [102 and 103].

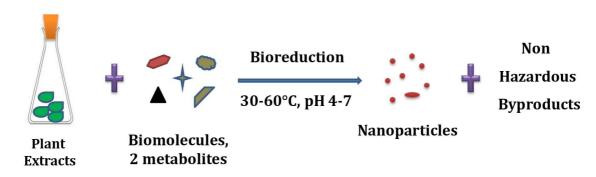


FIGURE3: PROPOSED PROTOCOL FOR SYNTHESIS OF NANOPARTICLES USING PLANT EXTRACTS {Recreated from [24]}.

A number of factors such as temperature, concentrations of the extract, pH and also the metal ions can affect the size and shape of the synthesized nanoparticles [104]. The plant extract based synthesis procedures usually have a high rate of reaction, taking several minutes to several hours for completion, depending on the type and amount of the plant extract. Most plants, especially the perennial plants, are almost always naturally available. Microorganism based nanosynthesis involves heating of the reaction mixture or culture medium, while in case of plant extract mediated synthesis of metallic nanoparticle, there is no need of heating, as it accomplished at room temperature . Plant extract based synthesis includes simple handling and easy reaction conditions, and therefore, it is having more suitability for large-scale production, when compared to microorganism based nanosynthesis [105, 106 and 107]. Table 1 shows various plant resource which are used to synthesize metal nanoparticles.

S.No.	Name of Plant/Plant Active Used	Metal Nanoparticles Made	Parts of Plant Used	Nanoparticle Size (Nm)	Nanoparticle Shapes	Pharmacological Activity	Reference
1.	Citrus maxima	Ag	Fruit	11.3–12.8 nm	Spherical	Antimicrobial (evaluated against Acidovorax oryzae strain RS-2)	[108]
2.	Telfairia occidentalis	AgO	Leaf	15.84 - 19.2 nm	Spherical	Antibacterial (evaluated against <i>K. Pneumonia</i>)	[109]
3.	Vitex negundo L.	Ag	Leaf	10 – 100 nm	Spherical	Antibacterial (evaluated against Proteus vulgaris, E. coli, P. aeruginosa, K. pneumonia, Salmonella paratyphi and S. aureus)	[110]
4.	Eucalyptus globulus	ZnO	Leaf	40 nm	Needle and Spherical	Antimicrobial (evaluated against , <i>S. aureus</i> ATCC 43300, <i>S. aureus</i> ATCC 25923, <i>E. faecalis</i> ATCC 29212, <i>E. coli</i> ATCC 25922, <i>S.</i> <i>enteritidis</i> ATCC 13076 and <i>P.</i> <i>aeruginosa</i> ATCC 27853, <i>S. typhimurium,</i> <i>K. pneumoniae,</i> , <i>Acinetobacter</i> <i>baumannii</i> and <i>Candida</i> <i>albicans</i>) and Anti- biofilm activity (evaluated against <i>P. aeruginosa</i> ATCC	[111]

TABLE1: PLANT RESOURCES BASED METAL NANOPARTICLES.

						27853 and S. aureus	
5.	Solanum nigurum	Au, Pd and Ag	Leaf	3.46 nm (Ag) 9.39 nm (Au) and 21.55 nm (Pd)	Spherical	ATCC 25923) Antibacterial, Antimicrobial (evaluated against <i>Escherichia coli</i>)	[112]
6.	Pimpinella anisum	Ag and Au	Seed	18-22 nm (Ag) 16-22 nm (Au)	Spherical	Antioxidant (evaluated by 1,1-diphenyl-2- picryl-hydrazyl (DPPH)), Antibacterial (evaluated against <i>S.</i> <i>aureus</i> and <i>E. coli</i>) and Antifungal (evaluated against Aspergillus flavus and Candida albicans)	[113]
7.	Aesculus hippocastanum	Ag	Leaf	50 ± 5 nm	Spherical	Antioxidant (evaluated by 1,1-diphenyl-2- picryl-hydrazyl (DPPH), Reducing power assay, Superoxide anion radical scavenging assay), Antibacterial (evaluated against Staphylococcus aureus, S. epidermidis, Listeria monocytogenes, Corynebacterium renale, Micrococcus luteus, Bacillus subtilis, B. cereus, Enterococcus faecalis, Pseudomonas aeruginosa, P. fluorescens, Escherichia coli, Enterobacter aerogenes, Klebsiella pneumonia, Proteus mirabilis, Candida albicans, C. tropicalis and C. krusei)	[114]
8.	Phyllanthus emblica	Ag	Fruit	19 - 45 nm	Hexagonal	Antibacterial (evaluated against Klebsiella pneumoniae and Staphylococcus aureus)	[115]
9.	<i>Allium</i> saralicum R.M. Fritsch	Zn	Leaf	~19 nm	Spherical	Anticancer (evaluated on HUVEC line by MTT assay), Antioxidant (evaluated by 1,1- diphenyl-2-picryl- hydrazyl (DPPH)), Cutaneous wound healing (evaluated on Sprague Dawley rats), Antimicrobial (evaluated against <i>C.</i> guilliermondii, <i>C. krusei</i> , <i>C. glabrata, C. albicans,</i> <i>Staphylococcus aureus</i> (ATCC No. 25923), <i>B. subtilis,</i> <i>E. coli</i> 0157:H7, <i>P.</i> <i>aeruginosa, Salmonella</i> <i>typhimurium</i> (ATCC No. 14028) and <i>Streptococcus</i> <i>pneumonia</i> (ATCC No. 49619))	[116]
10.	Averrhoa	ZnO	Fruit	35.4 - 59.5	Spherical	Antibacterial	[117]

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	bilimbi			nm		(evaluated against planktonic and biofilm <i>Escherichia coli</i>)	
11.	Curcuma longa L.	Ag	Leaf	15 - 40 nm	Spherical	Antimicrobial (evaluated against S.aureus, S.pyogenes, E.coli, P.aeruginosa, and C.albicans) Antibacterial	[118]
12.	Echinochloa frumentacea	ZnO	Grain	35 - 65 nm	Hexagonal	Antibacterial (evaluated against <i>S.</i> <i>typhi</i> and <i>B. pumilus</i>), Cytotoxicity study (evaluated on E.coli AB 1157)	[119]
13.	Bergenia ciliata	ZnO	Root	~30 nm.	Flower-Like	Antimicrobial (evaluated against Yersenia enterocolitica, S. aureus, S. typhi, P. aeruginosa,, E. coli, and Bacillus Subtilis), Anticancer (evaluated on HeLa cells of human cervical cancer and Human colon cancer (HT-29) cells by MTT assay), Antioxidant (evaluated by ABTS and 1,1- diphenyl-2-picryl- hydrazyl (DPPH))	[120]
14.	Crateva adansonii	ZnO	Leaf	30-55 nm	Hexagonal	Antimicrobial (evaluated against P. aeruginosa, B. subtilis, S. aureus, and E. coli)	[121]
15.	Hibiscus rosasinensis	Ag	Leaf	15–30 nm	Spherical	Antibacterial, Antimicrobial (evaluated against <i>S.</i> <i>aureus</i> and <i>E. coli</i>)	[122]
16.	Allium rotundum l., Ferulago angulate Boiss and Falcaria vulgaris Bernh.	Ag	Leaf	20.5 nm	Spherical	Antibacterial, Antimicrobial (evaluated against Pseudomonas aeruginosa PAO1 and Staphylococcus aureus ATCC 25923)	[123]
17.	Aegle marmelos	Ag	Fruit	159 - 181 nm	Spherical	Antimicrobial (evaluated against E. coli, Bacillus cereus, P. aeruginosa, Staphylococcus aureus, S. typhi, Shigella dysenteriae, Yersinia pestis)	[124]
18.	Artocarpus Hetrophyllus	Ag ₂ O	Leaf	~14 nm	Spherical	Antibacterial, Antimicrobial	[125]
19.	Azadirachta indica	Ag and ZnO	Leaf	8 - 50 nm	-	Antibacterial (evaluated against <i>Bacillus subtilis</i> (MTCC No.10619))	[126]
20.	Clitoria ternatea	Ag and Au	Flower	18-50 nm	-	Antibacterial (evaluated against <i>E.</i> <i>coli, Streptococcus</i> <i>pyogenes, K. pneumonia</i> <i>and S. aureus</i>), Antioxidant (evaluated by 1,1-diphenyl-2- picryl-hydrazyl (DPPH))	[127]
21.	Curcumin	SiO ₂	-	36 - 40 nm	-	Antimicrobial (evaluated against <i>P.</i>	[128]

						aeruginosa and S.	
22.	Calotropis gigantea L.	Ag	Flower	-	-	aureus) Antibacterial (evaluated against Bacillus subtilis, Pseudomonas putida and Escherichia coli)	[129]
23.	Terminalia belerica, Terminalia chebula, Swertia chiryta, Plumbago zeylanica, Holarrhena antidysenterica	CuO, ZnO, FeO	Leaf, Stem, Fruit, Root and Bark	2-10 nm (CuO), 2-10 nm (ZnO), 15- 23 nm (FeO)	Spherical (CuO). Spherical (ZnO), Spherical (FeO)	Antibacterial (evaluated against <i>Staphylococcus</i> aureus (ATCC-6538), Escherichia coli (ATCC- 8739) and <i>Salmonella</i> enteric (MTCC-3858))	[130]
24.	Myrtus communis L.	Ag	Leaf	~15 nm	Spherical	Antibacterial (evaluated against Methicillin- resistantStaphylococcus aureus(MRSA) ATCC 43300andEscherichia coliATCC 35218)	[131]
25.	Citrullis lanatus var	Au	Fruit	100-350 nm (Spherical), 200-500 nm (Triangular pyramid)	Spherical, Triangular pyramid	Antibacterial (evaluated against <i>S. epidermis</i> and <i>E. coli</i>)	[132]
26.	Ageratum conyzoides	Fe	Whole Plant	11.45-614.03 nm	Cubic	Antimicrobial (evaluated against Staphylococcus aureus (ATCC-25923), Bacillus subtilis, Pseudomonas aeruginosa (ATCC- 27853), C. albicans and Escherichia coli (ATCC- 25922)), Photocatalytic activity (Methylene blue degradation)	[133]
27.	P. domestica L.(Plum), P. Persia L. (Peach) and A. deliciosa (Kiwi)	TiO2	Fruits Peel	47.1-63.2 nm (Plum) 54.1-85.1 nm (Kiwi) 200 nm (Peach)	Cylindrical	Antibacterial (evaluated against <i>B. substilis, S.</i> <i>aureus, P. aeruginosa</i> and <i>E. coli</i>), Antioxidant (evaluated by 1,1-diphenyl-2- picryl-hydrazyl (DPPH), reducing power assays, hydrogen peroxide, and nitric oxide radical scavenging)	[134]
28.	Prosopis farcta	Ag	Fruit	10.26-14.65 nm	Spherical	Antioxidant (evaluated by 1,1-diphenyl-2- picryl-hydrazyl (DPPH)), Antimicrobial (evaluated against, <i>S.</i> <i>pneumonia, S. typhi, S.</i> <i>aureus</i> and <i>E. coli</i>)	[135]
29.	Camellia sinensis	Au and Ag	Leaf	~10 nm (Au), ~30 nm (Ag)	Spherical	Antimicrobial [evaluated against <i>S. aureus</i> (NBRC 12732) and <i>K. pneumoniae</i> (NBRC 13277)]	[136]
30.	Jatropha curcas	Ag	Seed	80 nm - 95 nm	Spherical	Antibacterial (evaluated against P. aeruginosa, B. subtilis and E. coli)	[137]
31.	Juglans regia	Ag	Leaf	20-30 nm	Spherical	Antibacterial (evaluated against K. pneumonia, S. aureus P. vulgaris, P. aeruginosa and E. coli)	[138]
32.	<i>Embelia ribes</i> Burm.f.	Ag	Fruit	~30 nm	Spherical	Antibacterial (evaluated against <i>Bacillus</i>	[139]

						subtilis), Anticancer (evaluated by MTT assay on MCF-7	
						cell lines)	
33.	Cardiospermum halicacabum	ZnO	Leaf	~48 nm	Cubic	Antibacterial (evaluated against, P. aeruginosa, S. saprophyticus, B. subtilis and E. coli)	[140]
34.	Panax ginseng	Zn	Root	45 nm - 85 nm	Spherical	Anticancer (evaluated by MTT assay on L20B tumor cell lines)	[141]
35.	Ocimum americanum L.	ZnO	Leaf	~21 nm	Spherical	Antimicrobial (evaluated against K. pneumonia, B. cereus, S. typhi, S. aureus, V. parahaemolyticus, Xanthomonas citri, P. aeruginosa, E. coli, Aspergillus parasiticus, C. albicans, Antioxidant (evaluated by 1,1-diphenyl-2- picryl-hydrazyl (DPPH))	[142]
36.	Terminalia arjuna	Au	Leaf	15 - 30 nm	Spherical	Antibacterial (evaluated against <i>S. typhimurium</i> , <i>P. aeruginosa</i> and <i>S.</i> <i>aureus</i>)	[143]
37.	Berberis vulgaris	Ag	Leaf and Root	30 - 70 nm	Spherical	Antibacterial (evaluated against <i>E. coli</i> and <i>S. Aureus</i>)	[144]
38.	Cressa cretica	Au	Leaf	15-22 nm	Spherical, pentagonal, rod and hexagonal	Antibacterial (evaluated against <i>S. aureus, S. pyogenes, K. pneumonia</i> and <i>E. coli</i>), Catalytic efficacy (4- nitrophenol reduction)	[145]
39.	Cassia siamea	ZnO	Leaf	below 100 nm	Slightly Ellipsoidal/ Spherical	Antimicrobial (evaluated against Pseudomonas aeruginosa, Staphylococcus saprophyticus, Streptococcus pyogenes and Proteus mirabilis)	[146]
40.	Hesperidin and Naringin	Ag and Au	-	100–225 nm	Spherical	Antibacterial (evaluated against neuropathogenic <i>E. coli</i> and methicillin resistant <i>S. aureus</i>)	[147]
41.	Citric Acid, Eugenol, Scopolamine, D- Glucose, Khelin Coumarin, Sucrose, Thymol and L- Asorbic	Au	-	30-80 nm (glucose), 150 nm (eugenol), 230 nm (thymol)	Spherical	Antibacterial, Anticancer, Antifungal	[148]
42.	Catha edulis Forsk (Khat)	CuO	Leaf	-	-	Antibacterial (evaluated against <i>S. typhimurium</i> and <i>E. coli</i>)	[149]
43.	Rhizophora apiculata	Ag	Leaf	-	-	Hepatoprotective (evaluated on hepatotoxin-induced liver damage in male Swiss albino mice)	[150]
44.	Annona reticulata	Ag	Leaf	6.48 - 8.13 nm	Cubic	Mosquito Larvicidal bioassay (evaluated on vector of dengue Aedes aegypti), Antibacterial (evaluated against E. coli, Bacillus	[151]

						cereus	
						S. aureus and P. aeruginosa)	
45.	Cleome viscosa	Ag	Fruit	20 - 50 nm	Spherical	Antibacterial (evaluated against <i>B. subtilis, K.</i> <i>Pneumonia, E. coli</i> and <i>S.</i> <i>aureus</i>), Anticancer (evaluated on PA1-0varian teratocarcinoma cell line and A549-Human lung adenocarcinoma by MTT assay)	[152]
46.	Carissa carandas	Ag	Fruit	10-60 nm	Spherical	Antibacterial (evaluated against Aeromonas hydrophila, Acinetobacter sp., and Staphylococcus aureus)	[153]
47.	Nyctanthes arbor-tristis	ZnO	Flower	12-32 nm	-	Antifungal (evaluated against Aspergillus niger, Penicillium expansum, Alternaria alternata, Botrytis cinerea and Fusarium oxysporum)	[154]
48.	Crocus sativus L.	Ag	Flower	12-20 nm	Spherical	Antibacterial (evaluated against Bacillus subtilis, E. coli, K. pneumonia, P. aeruginosa and Shigella flexneri)	[155]
49.	Ocimum Sanctum, Quercetin	Ag	Leaf	14.6 nm (Tulsi extract) 11.35 nm (Quercetin)	Spherical	Antibacterial (evaluated against Escherichia coli)	[156]
50.	Angelicae Pubescentis Radix	Ag and Au	Root	12.48 nm (Ag), 7.44 nm (Au)	Quasi- spherical(Ag), Spherical Icosahedral (Au)	Antioxidant (evaluated by 1,1-diphenyl-2- picryl-hydrazyl (DPPH), Antimicrobial (evaluated against <i>S.</i> <i>aureus, Salmonella</i> <i>enterica, E. coli</i> and <i>P.</i> <i>aeruginosa</i>)	[157]

CONCLUSION

A number of metals used for plant resources based green synthesis of nanoparticles like gold (Au), silver (Ag), zinc oxide (ZnO), copper (Cu), titanium (Ti), nickel (Ni), platinum (Pt), selenium (Se), cobalt (Co), palladium (Pd) and magnetite (Fe₃O₄) have already been reported, and have been proved as a potent remedy against different infectious diseases including other acute ailments [158, 24 and 159]. The plant mediated metal nanoparticles have been shown to possess various therapeutic activities such as antioxidant, antimicrobial, anti-inflammatory, anticancer, antidiabetic and immunomodulatory [160, 161, 162 and 163]. Some previous studies suggested that, various phytochemicals such as flavonoids, sugars, alkaloids, proteins, phenols and terpenoids are responsible for bioreduction, capping and stabilization of metal ions during nanoparticle synthesis [164 and 165].

Despite the ease involved in the purification of nanoparticles synthesized using only one single active substance in plant extract, it is important to further study the metal nanoparticles with a biomedical perspective for the treatment of particular diseases. At present, limited information is available in the scientific literature regarding the use of a single substance from plant extract for the synthesis of metal nanoparticles. Recent studies suggested that the flavonoids, due to its ample presence in the plant extracts, have a big contribution toward the bioreduction, capping and stabilization of metal ions into nanoparticle formation [156, 166, 167, and 168].

This exhaustive compilation work will be beneficial for researchers, as using plant sources for metal nanoparticle synthesis is energy efficient, cost effective, protecting human health and environment leading to lesser waste and safer products. Plant resources can act as both reducing and capping agent

during nanoparticle synthesis, and thus stabilizes the metal nanoparticles in shape and size controlled manner. This environment-friendly approach could be a competitive alternative to the traditional physical and chemical approaches used for synthesis of metal nanoparticle and thus has a potential to use in biomedical applications like in the field of cosmetics, medical devices, food industries, pharmaceuticals, opto-electronics and thus will definitely lead into a highly productive and innovative phase of research in near future.

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