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Synthesis and Antibacterial activity of Silver nanoparticles from *Prunus armeniaca* (Apricot) fruit peel extract

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ABSTRACT

Present study deals with synthesis of silver nanoparticles by using fruit peel extract of Prunus armeniaca (Apricot). Silver nanoparticles synthesized by simple, fast and ecofriendly method was characterized by scanning electron microscopy (SEM) and X-ray diffraction (XRD). Average size of silver nanoparticle was derived by Debye-Scherer's equation and found out to be 50nm. The synthesized silver nanoparticles were also evaluated for antibacterial activity against some microorganisms as zone of inhibition. Maximum zone of inhibition by silver nanoparticles was found against S. aureus which was close to reference drug.

Keywords: Fruit peel, silver nanoparticles, Prunus armeniaca, Antibacterial activity

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INTRODUCTION

Exploring the waste materials in different area of human applications has given much attention in recent years. Use of plant waste materials (e.g. fruit peel) for value added discoveries for identification of major biological potentials is one such area where scientists are focusing with zeal [1-4]. Synthesis of metal nanoparticles also one such area where plant materials such as peels, leafs, seeds and roots were extensively utilized in these days [5,6].Utilization of these waste materials in the synthesis of nanoparticles is studies under the light of green chemistry. Nanoparticles are regarded fundamental of nanotechnology. In recent past, noble metal nanoparticles have been main subject of research owing to their excellent electronic, optical, mechanical, magnetic and chemical properties [7]. Metal nanoparticles have immense applications in the area of diagnosis, biological and catalysis [6] Ag, Pt, Au and Zn are among major metal used in the formation of nanoparticles [8,9]. To the preparation of different metal nanoparticles for pharmaceutical applications, researchers adopting environmental friendly approach in these days [10]. Conventional techniques for the synthesis of nanoparticles such as chemical precipitation, sol-gel process, micelle, hydrothermal method, chemical vapor disposition etc. generate toxic chemicals that may causes adverse effects [11, 12]. Biological methods using plants parts extracts, microorganisms(bacteria, fungi, yeast) have been a valuable choice and regarded as best alternatives to chemical methods for the synthesis of metal nanoparticles [12-14]. Current bio-inspired methods are economical and less chances of use of toxic chemicals as well as high pressure and temperatures to the preparation of metal nanostructures [15,16].

In present research, we are going to report synthesis of silver nanoparticles by methanolic extract of *Prunus armeniaca* peel extract. Morphological characterization of present silver nanoparticles is characterized by X-ray diffraction (XRD) and scanning electron microscopy (SEM). Furthermore, silver nanoparticles were evaluated for antibacterial activity against selected bacterial pathogens.

MATERIALS AND METHODS

In present research chemicals were purchased from sigma Aldrich (USA) and no further purification was carried out. I R spectra were recorded by KBr discs using FT/IR - 4100 JASKO model in the ratio of 1:100. X-ray diffraction (XRD) patterns were recorded on a Rigaku Ultima IV X-ray Diffractometer at angles between 2° and 80° , with a scan rate of 2° /min.

Scanning electron microscopy was done using Carl Zeiss EVO LS10 (Oberkochen, Germany).

Coating was carried out using quorum Q150 R S (Quorem Technologies Ltd, 2 Acorn House, The Broyle, Ringmer, East, Sussex, United Kingdom) sputter coater by the deposition of gold on the samples.

Preparation of apricot extract: For the synthesis of silver nanoparticles rosacea fruit, apricot peel extract were used as reducing and capping agent.

About 1 kg of apricot fruit were purchased from local market. Peels of the fruit were dried and powdered. In an about 500mL beaker containing methanol powdered peel were dipped and kept overnight. After 24 hr, the methanolic extract was evaporated till syrupy consistency. Extract was kept in a refrigerator at 4° c for the synthesis of silver nanoparticles.

Synthesis of silver nanoparticle: An aqueous solution of silver nitrate (AgNO₃) at concentration of 0.02 mmol/L was prepared and used for the synthesis of silver nanoparticles. About 10 mL of apricot (*Prunus armeniaca*) peel extract was added into 80 mL of aqueous solution of 0.02 mmol/L silver nitrate (AgNO₃) for the reduction into Ag+ ions and exposed to sunlight and a color change takes place from yellowish to reddish brown after repeated experiments two or three times.

Evaluation of antibacterial activity: The silver nanoparticles synthesized from apricot (*Prunus armeniaca*) peel extract was evaluated for antibacterial activity by disc diffusion method against different pathogenic bacterial strains such as *E.coli, S.aureus, P.aeruginosa* and *Bacillus substilis*. A standard inoculum (1-2x10⁷ c.f.u/mL) 0.5 McFarland standards was inserted on the surface of sterile agar plates. A glass spreader was used for even distribution of the inoculum. The disc of 6.25mm were made from whatman no 1 filter paper and subsequently sterilized by dry heat at 140°C.

RESULTS AND DISCUSSION

X-Ray Diffraction Analysis

The x-ray diffraction pattern of the synthesized silver nanoparticles from extract of apricot peel was demonstrated by major peaks corresponding to 2θ values ranging from 20-50 as shown in fig 1. These peaks are supported by Joint Committee on Powder Diffraction Standards (JCPDS) file No.04-0783. Average size of silver nanoparticles was calculated by Debye-Scherer's formula and found out to be 50nm.

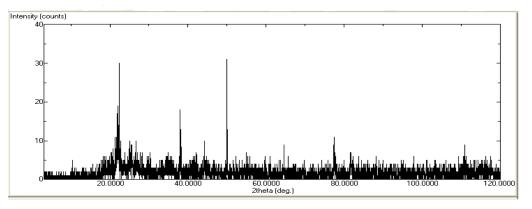


Fig 1. XRD pattern of silver nanoparticles

Scanning Electron Microscopy (SEM)

SEM analysis was done to identify the size and shape of silver nanoparticles obtained as a result of apricot peel extract with silver nitrate. Silver nanoparticles have somewhat rod shaped including some particles aggregates. Aggregates molecules appeared $20\mu m$ range as shown in fig 2.

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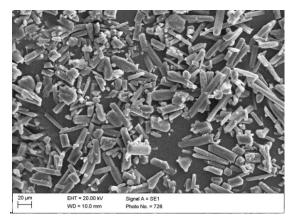


Fig 2. SEM images of silver nanoparticles

Antibacterial activity

Antibacterial activity assay of apricot peel extract mediated silver nanoparticles was performed against some selected bacterial pathogens by disc diffusion method. Antibacterial activity was compared among different bacterial strains along with standard ciprofloxacin as zone of inhibition (mm). Maximum and minimum zone of inhibition of synthesized silver nanoparticles was found against *S.aureus* (24mm) and *P.aeruginosa* (16mm). Standard drug (28mm) produce zone of inhibition 28mm. Silver nanoparticles shows good results against *S.aureus* (24mm) very close to standard drug. Zone of inhibition against two other strains (*E.coli, Bacillus substilis*) are 19&21 mm respectively (Table 1).

Table1:- Inhibition zone for silver nanoparticles synthesized by apricot (Prunus armeniaca) peel
extract against selected bacterial strains.

extract against selected bacterial strains.	
Pathogens	Zone of inhibition(mm)
E.coli	19
S.aureus	24
P.aeruginosa	16
Bacillus substilis	21
Ciprofloxacin(Std.)	28

Antibacterial activities of silver nitrate solution and nanoparticles have been reported in plenty of literatures [17]. Mechanism of antibacterial propensity of silver nanoparticles (AgNO₃Np) is still a matter of discussion. Reports reveals that, due to electrostatic interaction between negatively charged cell membrane of microorganism and positively charged of Ag+ ions of nanoparticles is primarily responsible for antibacterial activity. Some researchers reported that generation of free radicals by silver nanoparticles are responsible for damaging the membrane of pathogens [18].

CONCLUSION

In conclusion, in present research we have prepared silver nanoparticles from apricot fruit peel extract and characterized by x-ray diffraction and scanning electron microscopy. Fully characterized nanoparticles were evaluated against some bacterial pathogens. Antibacterial activity of synthesized nanoparticles finds good results as compared to standard drug. Thus, production of such nanoparticles is worthy in terms of economy and also environmentally viable approach. We will try few more metal nanoparticles by selecting other fruit peels for different biological activities. Toxicity studies of *Prunus armeniaca* peel mediated silver nanoparticles are progress in our laboratory.

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